

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Subsidiary Level and Advanced Level

CHEMISTRY

Advanced Practical Skills

CONFIDENTIAL INSTRUCTIONS

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.



The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.

If you have any problems or queries regarding these instructions, please contact CIEby e-mail:international@cie.org.ukby phone:+44 1223 553554 by fax:+44 1223 553558stating the Centre number, the nature of the query and the syllabus number quoted above.

This document consists of 8 printed pages.

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UNIVERSITY of CAMBRIDGE International Examinations

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Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution. Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers, safety goggles and plastic gloves.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn in particular, to certain materials used in the examination. The following codes are used where relevant.

- C corrosive substance F highly flammable substance
- Hharmful or irritating substanceOoxidising substance
- T toxic substance N dangerous for the environment

The attention of Supervisors is drawn to any local regulations relating to safety, first-aid and disposal of chemicals.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

Before the Examination

1 Access to the question paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate to within one part in two hundred of those specified.

3 Labelling of materials

Materials must be labelled as specified in these instructions. Materials with an **FA** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FA** coded chemical is given in the question paper.

4 Identity of materials

It should be noted that descriptions of solutions given in the question paper may not correspond exactly with the specifications in these instructions. The candidates must assume the descriptions given in the question paper.

5 Size of group

In view of the difficulty in preparing large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), safety goggles and disposable gloves should be used where necessary.
- **3** For each candidate
 - $1 \times \text{glass rod}$
 - $1 \times 250 \, \text{cm}^3$ beaker
 - $2 \times 250 \, \text{cm}^3$ conical flask
 - $1 \times 50 \, \text{cm}^3$ measuring cylinder
 - $1 \times 250 \, \text{cm}^3$ graduated (volumetric) flask
 - 3 × teat/squeeze/dropping pipette
 - $1 \times 50 \, \text{cm}^3$ burette
 - $1 \times \text{stand}$ and burette clamp
 - $1 \times \text{small funnel for filling burette}$
 - $1 \times 25 \text{ cm}^3$ bulb form pipette
 - 1 × pipette filler
 - 1 × small funnel
 - 1 × white tile
 - 1 × heat proof mat
 - 1 × Bunsen burner
 - 1 × tripod
 - 1 × pipe-clay triangle
 - $1 \times \text{crucible}$ (at least 15 cm^3 capacity)
 - $1 \times pair of crucible tongs$
 - $1 \times spatula$
 - 3 × boiling tube*
 - 6 × test-tube*
 - 1 × test-tube holder
 - 1 × test-tube rack
 - $1 \times$ wash bottle containing distilled water
 - paper towels
 - access to a balance weighing to 0.1 g or better
- * Candidates are expected to rinse and re-use test-tubes and boiling tubes where possible. Additional tubes should be available.

Where access to a balance is limited candidates should be directed to start the practical examination on different questions. (See p62 of the 2011 Syllabus for balance: candidate ratio).

Chemicals Required

It is especially important that great care is taken that the confidential information given below does **not** reach the candidates either directly or indirectly.

2 Particular requirements

notes (hazards given in this column refer to the raw materials) 5.80–5.90g of (NH ₄) ₂ SO ₄ .FeSO ₄ .6H ₂ O [H] should be finely ground provided in a stoppered tube. Cautiously pour 55 cm ³ of concentrated (98%) sulfuric acid [C] into 500 cm ³ of distilled water with continuous stirring. Make the solutior to 1 dm ³ with distilled water. Care: Concentrated H ₂ SO ₄ <i>is very corrosive</i> . Dissolve 1.58g of KMnO ₄ [N] [O] [H] in each dm ³ of distilled water. Dissolve 1.58g of KMnO ₄ [N] [O] [H] in each dm ³ of distilled water. a finely ground mixture of 9 parts by mass hydrated magnesium su MgSO ₄ .7H ₂ O, to 1 part by mass anhydrous sodium sulfate, Na ₂ SO finely ground NAH ₄ Br provided in a stoppered tube finely ground NH ₄ Br provided in a stoppered tube	identity ammonium iron(II) sulfate ammonium iron(II) sulfate 1 moldm ⁻³ sulfuric acid 0.010 moldm ⁻³ potassium manganate (VII) 9 parts MgSO ₄ .7H ₂ O to 1 part anhydrous Na ₂ SO ₄ by mass sodium hydrogencarbonate ammonium bromide	per candidate 5.80- 5.90g 5.90g 150cm ³ 1.5g 2.5g 1.5g 1.5g 1.5g 1.5g	label FA 1 FA 2 FA 3 FA 6 FA 6 FA 6	Junction Junction Image: Constraint of the state of the stat
finely ground NaHCO $_3$ provided in a stoppered tube finely ground NH $_4$ Br provided in a stoppered tube	sodium hydrogencarbonate ammonium bromide	2.5g 1.5g	FA 5 FA 6	
a finely ground mixture of 9 parts by mass hydrated magnesium sulfate, $\rm MgSO_4.7H_2O,$ to 1 part by mass anhydrous sodium sulfate, $\rm Na_2SO_4$	9 parts MgSO ₄ .7H ₂ O to 1 part anhydrous Na ₂ SO ₄ by mass	1.5g	FA 4	
Dissolve 1.58g of KMnO $_4$ [N] [O] [H] in each dm 3 of distilled water.	0.010 moldm ⁻³ potassium manganate (VII)	150 cm ³	FA 3	۲ Σ
Cautiously pour 55 cm ³ of concentrated (98%) sulfuric acid [C] into 500 cm ³ of distilled water with continuous stirring. Make the solution up to 1 dm ³ with distilled water. <i>Care: Concentrated</i> H_2SO_4 <i>is very corrosive</i> .	1 moldm ⁻³ sulfuric acid	250 cm ³	FA 2	Ε
5.80–5.90 g of (NH ₄) ₂ SO ₄ .FeSO ₄ .6H ₂ O [H] should be finely ground and provided in a stoppered tube.	ammonium iron(II) sulfate	5.80- 5.90g	FA 1	Ξ
notes (hazards given in this column refer to the raw materials)	identity	per candidate	label	azard

NOTE: The laboratory must be well ventilated

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hazard	ard label	per candidate	notes (hazards given in this column refer to the raw materials)
<u>ເ</u>] concentrated sulfuric acid	1 cm ³	Care: Concentrated H_2SO_4 is very corrosive.
[N] [J]	N] 0.1 moldm ⁻³ potassium chromate(VI)	10 cm ³	Dissolve 19.4 g of K_2 CrO ₄ [T] [N] in each dm ³ of solution.
Ξ] dilute hydrochloric acid		
ට] dilute nitric acid	I	
Ξ] dilute sulfuric acid		
Ξ] aqueous ammonia		
5] aqueous sodium hydroxide		
E] 0.1 mol dm ⁻³ barium chloride	See iden	ntity details and preparation instructions on page 65 and 66 of the 2011 syllabus
Ξ	0.1 mol dm ⁻³		
[N] [T]	N] 0.1 mol dm ⁻³ lead(II) nitrate		
[H] [N]	[N] 0.05 mol dm ⁻³ silver nitrate		
Ξ] limewater		
[N] [L]	N] acidifided aqueous potassium dichromate(VI)		

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Responsibilities of the Supervisor during the Examination

1 The Supervisor, or other competent chemist **must**, **out of sight of the candidates**, **carry out the experiment in Question 1 and Question 2** and complete tables of readings on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

This should be done for: each session held and each laboratory used in that session, and each set of solutions supplied.

N.B. The question paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the Examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the bar code label provided.
- 2 A copy of the Supervisor's Report relevant to the candidates in **1**.
- **3** A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.

5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

COLOUR-BLINDNESS

With regard to colour-blindness – a minor handicap, relatively common in males – it is permissible to advise candidates to request assistance on colours of, for example precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application for this handicap.

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REPORT FORM

This form must be completed and sent to the Examiner in the envelope with the scripts.

Centre Number Name of Centre

1 Supervisor's Results

Please submit details of the readings obtained in **Question 1** and **Question 2** on a spare copy of the question paper clearly marked 'Supervisor's Results' **and showing the Centre number and appropriate session/laboratory number**.

2 The candidate numbers of candidates attending each session were:

First Session

Second Session

- **3** The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:
 - (a) any general difficulties encountered in making preparation;
 - (b) difficulties due to faulty apparatus or materials;
 - (c) accidents to apparatus or materials;
 - (d) assistance with respect to colour-blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported direct to CIE on the normal 'Application for Special Consideration' form.

A plan of work benches, giving details by candidate numbers of the places occupied by the candidates for each experiment for each session, must be enclosed with the scripts.

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Report on any difficulties experienced by candidates.

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